



Name _____

Periodic Table Families Word Search

T R A N S I T I O N P N J B U U P D D L
 T F L C L C H M E Y M G T I K M J U A C
 F M K M T D W Y D E H M H J O G C T G D
 N E A L A I O M R J F D N Y H X E E A Z
 H T L I L A N T H A N I D E S M P D Z H
 V A I B N H U I V E V K F P N E M E X B
 J L F N R G Q O D C X T K O F W K H H J
 N O B L E G R N M E C E N S G Q Y H R V
 E K S H A R U O V T S Y N T H E T I C H
 Z Q C C C A T B U B N K W - L K O X L D
 R R K A T R L K X P Y M E T A L L O I D
 S X N A I C K B Q A H T E R L M B C S G
 C W G H V H Y O Q D K S H A L O G E N Q
 N F E D E I D O N H L K U N O P R Z G F
 I Y Y F H B S K P O L M H S Q Y J U N Z
 H D X O F Y R L A L K A L I N E M T S H
 P K Y C U R I X C L Y N Z T B U D T V E
 V D Z Y B K K T G Q P G Q I X H W B J O
 D S L P W V F J O E E N S O Q P X D P L
 W R K X B H N E H Q S E S N U F S L J N

A small periodic table showing element symbols and names. The table is organized into groups and periods. The groups are labeled with Roman numerals I through 10, and the periods are labeled with numbers 1 through 7. The elements are arranged in a grid, with the noble gases (He, Ne, Ar, Kr, Xe, Rn) in the far right column. The lanthanides and actinides are shown in separate rows at the bottom of the table.

- Alkali
- Alkaline
- Transition
- Lanthanide
- Actinide
- Halogen
- Noble
- Main-group
- Metal
- Nonmetal
- Metalloid
- Post-transition
- Synthetic
- Reactive
- Inert